

Assessment of Lead and Zinc Profile from Vehicle Emission in Roadside Fodder Plants

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Abstract: Due to increasing population and increasing number of automobiles the fodder samples were collected from Khushab District along the roadside to examine the heavy metal contamination. It was a serious issue because animals consume these contaminated fodders and then they enters the food chain where they cause toxicity. Five fodder plants *Pennisetum glaucum*, *Trifolium alexandrinum*, *Saccharum officinarum*, *Zea mays* and *Brassica compestris* were selected for sampling. The sampling of selected fodder plants and their soil was done during November which was the driest season of Khushab city and the dust and smoke was present on the fodder plants. The samples were air and oven dried. Then subjected to digestion. Acid digestion was done with Nitric acid (HNO_3), Hydrogen peroxide (H_2O_2) and Hydrogen chloride (HCL). The samples become transparent and the heavy metal analysis of both soil and fodder samples was done through Atomic absorption spectrophotometer (AASP). Lead and zinc heavy metals were analyzed. The pollution load index was lower than 1 which indicates that there was low concentration of metal in soil of the collected samples of the current study. It was obtained that the soil was less polluted. The daily intake of metals was lower than 1 that concluded that they were at safe limit during consumption. The present study showed the enrichment concentration of metals greater than 1 it means these metals were present in both soil and fodder samples growing near roadside.

Keywords: Heavy metals, Roadside, Fodder Plants, Daily intake

1. Introduction

The word pollution is derived from Latin word “Pollutioneum” meaning to make dirty. Atmospheric pollution can be defines as change in the air constituent due to human activity [1]. According to National Environmental Research Council, pollution can be defined as, release of wasteful substances and energy through human activities that cause harmful changes in the natural environment. Health Effects Institute reported that 95% of world’s population breathe dangerous and polluted air. The presence of one or more contaminants such as dust, smoke, smog that cause injurious effects on human health, plants and animal life, or which interferes with enjoyment of life and property is called Air pollution. The atmospheric pollution is the serious problem of today. It is the more severe type of pollution, it causes declines in environmental conditions. According to this report Pakistan, Bangladesh, India have contributing steepest in increasing air pollution level since 2010. In Pakistan tremendous increase in motor vehicles is 37% per year. Vehicles emit 20-90% of cadmium that have negative impacts on green pigment in plants [2].

Plants are very important to enhance the air quality by absorbing gases and particles from the air especially in urban areas [3]. The effective indicator of overall air pollution are plants. Plants are mostly used for the determination of air pollutants in a very short time. So that, the trees and Shrubs are used as dust filters to determine the increasing heavy metal and dust pollution level of atmosphere [4].

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Increased road transportation and industrialization cause the release of high concentration of heavy metals and harmful gases [5]. Air pollution have negative effects on environment and living organisms worldwide. Heavy traffic cause harmful effects on fodder crops near roadside area [6]. Air pollutants enter in to the plants through leaves and soil. Due to exposure to the air, harmful particles present in the air, when enters to these leaves cause changes to their physiology [7]. Industries and vehicles are accountable for emissions of air particles and harmful gases and particles are pointers of air pollution and fodder plants are very delicate to them [5].

Air pollutants when emitted cause deleterious effects on human, plants and also on environment [8-9]. Roadside plants are major victims of these air pollutants and thus have negative effects like they stop growth, damaging of leaves veins, maturity level becomes delayed, premature leave aging, early abortion of flower, and also effect quality and yield of plant [10-11].

In humans, increased level of these air pollutants cause severe and long lasting effects e.g. Respiration diseases early maturity, tumor of lungs, long lasting lung and heart diseases in humans. Many investigators study about the response of plants to these pollutants and also their effects on their structure, function and biochemistry. In the environment all plant species have a various impacts on different pollution type depending upon the ecological conditions. It is assumed that automobile pollution is more dangerous than any other type of pollution because humans mostly stay near automotive exhaust. Thus, through respiration these particles are inhaled directly [12-16].

In delicate class of plant species the automobile emissions cause damage to the leaf, stomata that leads to the early aging and reduce photosynthesis [17]. When the leaf damaged by air pollutants leaf size and number decreases and hence the radiations cannot enter into the leaves so the photosynthesis reduced or stops. When plants faced the polluted environment leaf surface area and petiole length reduced [17-18]. Many previous research workers also put forward the comparison of effect of polluted and non-polluted environment [19]. This study is planned to assess the heavy metal profile in soil and fodder plants contaminated with vehicular exhaust near roadside area.

2. Materials and methods

2.1. Study area

Khushab District

Khushab city is found in Punjab Pakistan. It is located at 32.30 latitude and 72.35 longitude. Its situation is 186 meters above the sea level. It is 38th biggest city in Punjab province. In Punjab province it is located between cities of Sargodha and Mianwali. The word Khushab means sweet water. It consists of agricultural lowlands, hills and lakes. The district has four Tehsils, Khushab, Noor PurThal, Qaidabad and Naushera. Its population was 110,868 at the time of 2009. The area code of Khushab is 0454. The increase in population occurs day by day so, the traffic density also become increased to large extent on the roads. The current study based on the heavy metal analysis coming from the vehicular exhaust along roadside fodder plants. Livestock farmers are mainly dependent on the green fodders grown in this area.

2.2. Collection of samples

Forage samples

Total 120 samples of five fodders *Pennisetumgluacum*, *Zea mays*, *Brassica compestris*, *Trifolium-alexanderium*, *Saccharumofficinarum* (Bajra, Makai, Sarson, Barseem and Gunna) respectively were collected from four sites of Sargodha to Khushab road (Table 1). Two samples of each fodder plant were collected, one from the edge of the road and one away from the road. Two replicates of each of above two samples were collected. Fodder samples were collected in polythene bags.

Soil Samples

The samples of soil were collected from the four sites of sampling near road from Sargodha to Khushab. 120 replicates of soil of each plant were collected. Each of these (1kg sample of soil) were



collected in polythene bags by digging soil about 15-30 cm deep with the help of shovel. The 0-25 cm depth was considered to represent the plough layer and average root zone for nutrients uptake and heavy metals burden by plants.

Table 1. List of Fodders

Sr. No	Common Name	Scientific name	Family
1	Barseem	<i>Trifolium alexanderium</i>	Fabaceae
2	Corn	<i>Zea mays</i>	Poaceae
3	Sarson	<i>Brassica compestris</i>	Brassicaceae
4	Pearl millet	<i>Pennisetum gluacum</i>	Poaceae
5	Sugarcane	<i>Saccharum officinarum</i>	Poaceae

2.3. Sample Preparation

Preparation of Fodder samples

Plants were first placed in open air and then transferred to oven at 72°C till all the moisture contents are removed. Then these dried samples were converted into powder form and stored in plastic bags until used for digestion. Wet digestion method was used for the digestion of fodder samples. Samples were digested in 5mL of 68% HNO₃ for 24 h. Then put into digestion chamber and heat it and removed until when the evaporation of fumes stops. Then 5mL of 30% H₂O₂ were added and again heated until it becomes colorless. By adding the filtered distilled water the digested samples were made up to 50 mL and kept in labeled plastic bottles [20]. The solution was then taken to atomic absorption Spectrophotometer for heavy metal analysis. The sample preparation procedure was performed according to [20].

Preparation of Soil samples

Soil samples was air dried at room temperature and then oven dried at 70°C. 0.5g of soil samples were subjected to digestion chamber. 0.5g of sample and 20mL of HNO₃ was added in digestion tubes. Heat the tubes in digestion chamber. Add 5mL of H₂O₂ was added and then heated again until it becomes colorless [20]. By adding distilled water the samples were made up to 50mL. Then the samples were stored in plastic bottles. These solutions were then subjected to Atomic absorption spectrophotometer for analysis of heavy metals.

2.4. Standard Preparation:

The standards were prepared by using following methods

- To make the standard solution glass were used and it must be clean. 100mL volumetric flask and funnel were washed with de-ionized water.
- Analytic balance was used to weight the sample in a beaker.
- Small volume of sample or another solvent can be used to dissolve the sample
- The sample was completely dissolved in solution. To speed the process of dissolution hot plate was used.
- Volumetric flask was used to pour the sample quantitatively. Beaker, funnel and stirring rod was washed completely with de-ionized water. Sample was transferred to the flask.
- The beaker was washed with de-ionized water for few minutes
- Made a mark below 100mL on volumetric flask and then filled it with few mL.
- By using dropper add few drops of de-ionized water so that meniscus touched the mark that was placed on the flask.
- Shake the flask for few minutes to mix it thoroughly
- Make sure that solution becomes homogeneous
- At the end again checked that meniscus touch the mark of 100mL at volumetric flask

2.5. Instrumentation

Table 2. Parameters of Instrument used for detection of heavy metals

Element	Lamp current (mA)	Wavelength (nm)	Slit Width(nm)	Instrument detection(mg/l)
Cd	4	228.8	0.5	0.0020
Fe	5	248.3	0.2	0.0060
Mn	5	279.5	0.2	0.0020
Zn	5	213.9	1.0	0.0010
Pb	5	217.0	1.0	0.0010
Ni	4	232.0	0.2	0.0010
Cr	7	357.9	0.2	0.0010
As	10	193.7	0.5	0.0010

Mineral analysis:

Plants and soil samples were directed towards the Atomic Absorption spectrophotometer for heavy metal analysis. The Atomic absorption spectrophotometer Perkin-Elmer AAS-5000 (Perkin-Elmer Corp, 1980), was used after acid digestion (Table 2). It was used to analyze the heavy metals lead (Pb) and Zinc (Zn) in the present study. As was analyzed through using graphite furnace.

2.6. Statistical analysis

SPSS (Special Programme for social sciences) was used for the analysis of variance and correlations. Variance analysis of heavy metals in soil and fodder plants was determined by two way ANOVA. The soil to fodder correlation was determined. Significance of mean values was at 0.05, 0.001, 0.01 probability levels reported by [21].

2.7. Bio-Concentration Factor (BCF)

The concentration of heavy metal in fodder sample from soil to fodder is called as bio concentration factor (on the basis of dry weight).

$$\text{Bio-concentration factor} = C_{\text{fodder}} / C_{\text{soil}}$$

C_{fodder} indicates the metal concentration present in the fodder sample while the C_{soil} indicates the metal concentration present in the soil, mg/kg dry weight [22].

2.8. Daily Intake of Metals (DIM)

The formula to determine the daily intake of metal was given by [23].

$$\text{DIM} = C_{\text{metal}} \times C_{\text{daily food intake}} \times \text{Conversion factor} / B_{\text{average weight}}$$

C_{metal} indicates the concentration of metals, $C_{\text{daily food intake}}$ indicates the consumption of fodder by cattle per day in Kg, the value of Conversion factor was 0.085, the $B_{\text{average weight}}$ represents the average body weight of buffalo of the selected sampling site. The average consumption of fodder by animal was 12.5Kg and average body weight was 550Kg per cattle [24].

2.9. Health Risk Index (HRI)

To determine the health risk of heavy metals through ingestion of contaminated fodders health risk index was calculated. This can be calculated by dividing the daily intake of metal to oral reference dose [25].

$$\text{HRI} = \text{DIM} / R_{\text{fD}}$$

DIM indicates the daily intake of metal while R_{fD} indicates the oral reference dose of animal. The oral reference dose of metals was

2.10. Enrichment Factor: (EF)

It can be defined as the withholding of metal in soil and it depends upon the availability of metal in soil. It was used to find the concentration of metal present in the soil.

$$EF = \frac{\text{concentration of metals in the amended soil}}{\text{concentration of metals in the controlled soil}} [26]$$

Pollution Load Index

The pollution load index helps us to determine the concentration of heavy metal in the soil.

$$PLI = C_{\text{soil}} / C_{\text{reference value}}$$

C_{soil} denotes investigated concentration of metal in the soil and $C_{\text{reference value}}$ indicates the reference value of soil for each metal [27].

2.11. Correlation

The correlation coefficient of metal concentration in soil and fodder samples were determined by using (SPSS) abbreviated as Statistical program for social sciences. To determine the correlation of metal from soil to fodder one way ANOVA was used in the current study.

3. Results and discussions

3.1. Soil

The analysis of variance of Pb in soil showed significant effect on Sites, Treatment, Fodders, Sites * Treatment, Sites * Fodders, Treatment * Fodders, Sites * Treatment * Fodders (Table 3). The significant effect was showed in analysis of variance of data for Zn Sites, Treatment, Fodders, Sites * Treatment, Sites * Fodders, Treatment * Fodders, Sites * Treatment * Fodders (Table 4).

Table 3. Analysis of variance of data for Pb in soil from four road sites

SOV	Degree of freedom	Mean square
Sites	3	129.957***
Treatment	1	47.066***
Fodders	4	21.155***
Sites * Treatment	3	19.363***
Sites * Fodders	12	24.215***
Treatment * Fodders	4	28.019***
Sites * Treatment * Fodders	12	47.259***
Error	80	0.217

Table 2. Analysis of variance of data for Zn in soil from four road sites

SOV	Degree of freedom	Mean square
Sites	3	277.733***
Treatment	1	570.946***
Fodders	4	71.483***
Sites * Treatment	3	65.981***
Sites * Fodders	12	84.486***
Treatment * Fodders	4	96.247***
Sites * Treatment * Fodders	12	78.061***
Error	80	76.915

3.2. Fodder

The analysis of variance of data for Pb showed significant effect on Sites, Sites* Treatment, and significant effect on Treatment, Sites* Fodders and non-significant effect on Fodders, Treatment * Fodders, Treatment* Fodders and Sites * Treatment * Fodders. The analysis of variance for Zn showed significant effect on Sites, Fodders, Sites * Treatment, Sites * Fodders, Treatment * Fodders, Sites * Treatment * Fodders while non-significant effect on Treatment (Table 6).

Table 3. Analysis of variance of data for Pb in fodder plants from four road sites

SOV	Degree of freedom	Mean square
Sites	3	288.280***
Treatment	1	24.377*
Fodders	4	2.820 ^{ns}
Sites * Treatment	3	116.857***
Sites * Fodders	12	8.785*
Treatment * Fodders	4	4.141 ^{ns}
Sites * Treatment * Fodders	12	3.659 ^{ns}
Error	80	4.025

Table 4. Analysis of variance of data for Zn in fodders from four sites having high vehicular exhaust

SOV	Degree of freedom	Mean square
Sites	3	233.774***
Treatment	1	1.253 ^{ns}
Fodders	4	2.358***
Sites * Treatment	3	66.136***
Sites * Fodders	12	7.983***
Treatment * Fodders	4	16.118***
Sites * Treatment * Fodders	12	5.261***
Error	80	.387

3.3. Mean concentration of heavy metals in fodder Plants

The highest and lowest mean concentration of Pb showed that the *B.compestris* was facing highest contamination of Pb through vehicles near and even at a distance from the road (Table 7). The highest concentration of Zn was present in *B.compestris* of site 1 and lowest (Figure 1) value was present in *S.officinarum* of site 2 present at some distance from the road (Table 8, Figure 2). The present study showed that the mean concentration Pb, Zn of fodder samples collected from roadside were higher as compared to the mean concentration given by [28]. The work done by [28]. The lead concentration was 0.0520mg/kg which was lower than the present study. Their work was done along major roads and control sites of Dibete Area. He also analyzed the heavy metal concentration along roadside forage samples. In Nigeria the same work was done by Ogandele, he collected forage sample along major roadsides. The samples were acid digested and analysis of metals were done by Atomic Absorption Spectrophotometer. According to present study range of Pb was 13.0352 mg/kg to 2.2152 mg/kg, which was within the range (24 to 397mg/kg) as proposed by [29]. The mean concentration of Ni and Zn were higher as compared to present study while the concentration of Cd and Pb was lower [30]. According to Ahmad [31], the levels of heavy metals concentration for Pb was lower in their forage samples as compared to the current study.

Table 5. Mean concentration of Pb in fodders from four sites contaminated with vehicular exhaust

Sites	Treatments	<i>P.glaucum</i>	<i>T.alexandrinum</i>	<i>S.officinarum</i>	<i>Z.mays</i>	<i>B.compestris</i>
Site 1	AR	7.0562	7.5520	6.8850	4.3330	6.8410
	NR	6.2162	6.4787	7.5845	7.4205	5.1725
Site 2	AR	4.4382	5.2496	3.6305	3.3047	4.7744
	NR	4.9105	2.2265	4.6580	4.9053	2.2152
Site 3	AR	2.2907	3.0527	3.0847	11.8012	8.8365
	NR	9.5643	9.7587	9.5032	2.4895	13.0325
Site 4	AR	11.7597	13.0973	11.5423	10.6847	10.8030
	NR	11.9575	10.1842	10.7693	12.1300	12.0213

Table 6. Mean concentration of Zn in fodders from five sites contaminated with vehicular exhaust

Sites	Treatments	<i>P.glaucum</i>	<i>T.alexandrinum</i>	<i>S.officinatum</i>	<i>Z.mays</i>	<i>B.compestris</i>
Site 1	AR	10.5592	10.7720	10.6618	4.2010	11.0570
	NR	10.6642	10.2452	10.7960	10.2763	4.6690
Site 2	AR	4.2017	4.7192	1.5527	2.1125	3.6005
	NR	5.0686	2.2995	3.8772	4.5483	1.8845
Site 3	AR	2.0128	1.7133	1.9630	8.1217	6.2375
	NR	7.3767	6.5393	6.6267	2.3454	6.0813
Site 4	AR	6.7621	6.2918	8.7850	7.8785	8.5588
	NR	8.5633	9.0408	8.2085	8.8212	8.7685

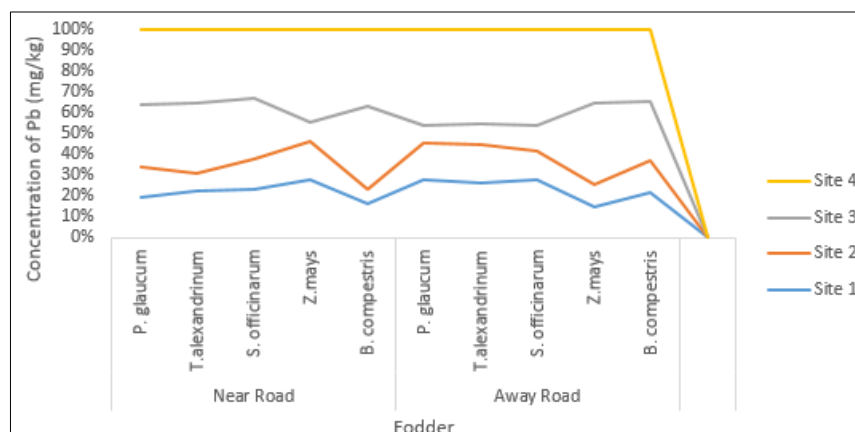


Figure 1. Pb fluctuations in fodders from four road sites of sampling

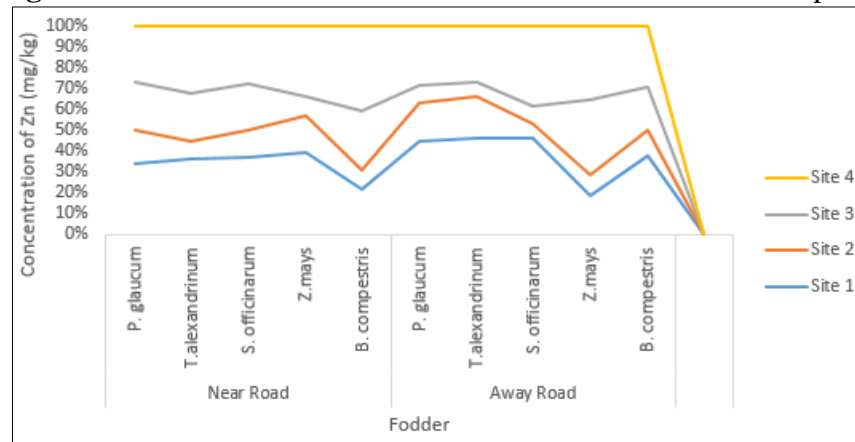


Figure 2. Zn fluctuations in fodders from four road sites of sampling

3.4. Mean concentration of heavy metals in soil

The soil of *P.glaucum* had highest concentration of Pb while the *Z.mays* showed the lowest concentration facing high traffic density (Table 9, Figure 3). The Zn showed highest and lowest concentration in the samples *B.compestris* and *Z. mays* fodder samples facing highly traffic pollutants (Table 10, Figure 4). The mean concentration of the zinc is lower as compared to present study [30]. The trend of heavy metal concentration of soil taken roadsides was given as Fe > Zn > Mn > Pb > Cd [32]. The concentration of metals were lower as compared to the present study. The higher Zn and Cd level along roadside soil samples were due to high traffic density and lubricating oils which contains Zn in the form of Zn dithiophosphates [33]. Opaluwa [34] reported the mean concentration of Fe, As, Pb, Zn, Ni was lower as compared to current study. Mmolawa [35] worked on soil samples along major roadsides of Botswana. He selected soil samples from five sites and analyzed the heavy metals in them.

The mean concentration given by Mmolawa showed lower values of Fe, Mn, Ni, and Zn as compared to the current study.

Table 7. Mean Concentration of Pb in Soil at various distances along the road

Sites	Treatments	<i>P. glaucum</i>	<i>T. alexandrinum</i>	<i>S. officinarum</i>	<i>Z. mays</i>	<i>B. compestris</i>
Site 1	AR	6.2360	7.3517	6.5338	6.6487	7.4677
	NR	8.1152	6.2293	5.8748	7.2912	4.0228
Site 2	AR	4.1487	5.6590	4.2318	4.4065	5.7683
	NR	4.8030	9.8523	8.4347	8.1777	8.0042
Site 3	AR	8.4307	10.1848	10.3493	8.5850	10.0333
	NR	13.2235	10.4850	11.5438	11.6182	11.2957
Site 4	AR	11.0618	11.9107	2.0442	2.0055	3.5673
	NR	3.7783	2.0832	3.5795	3.5450	19.7183

Table 8. Mean Concentration of Zn in Soil along the road from four sampling sites

Sites	Treatments	<i>P. glaucum</i>	<i>T. alexandrinum</i>	<i>S. officinarum</i>	<i>Z. mays</i>	<i>B. compestris</i>
Site 1	AR	10.0003	9.8918	10.1072	11.5793	11.4250
	NR	11.2262	11.6983	3.9968	11.4345	40.8750
Site 2	AR	3.3360	3.7797	3.5550	4.3617	4.0118
	NR	4.6643	7.1578	6.6658	7.2000	5.7330
Site 3	AR	6.1152	6.8988	8.5543	7.0552	7.0058
	NR	8.9343	7.6922	8.9680	8.5663	7.9033
Site 4	AR	8.7508	8.0092	1.9237	2.2705	2.1903
	NR	1.9030	1.8818	1.7315	1.7170	1.7448

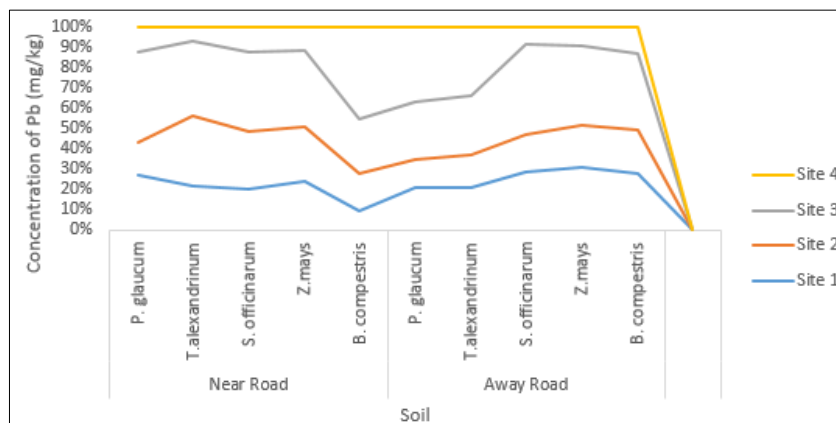


Figure 3. Pb fluctuations in Soil from four road sites of sampling

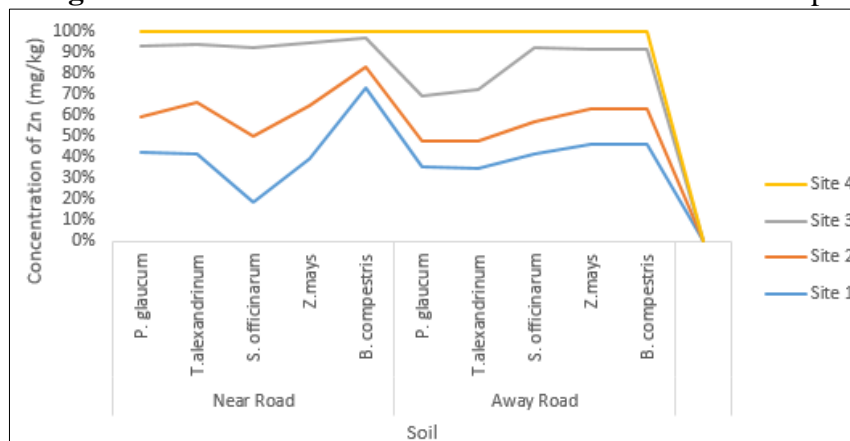


Figure 4. Zn fluctuations in Soil from four road sites of sampling

3.8. Bio concentration Factor

The highest mean concentration of Bio concentration factor was 5.646365mg/kg in *S.officinarum* at a distance from the road while lowest in *Z.mays* 0.214276mg/kg present near the road edge (Table 11). The Highest bio concentration of Zn was present in *Z.mays* samples of near roadside while lowest in *S.officinarum* samples collected away from the road (Table 12). The order of Bio concentration factor of the present study was Pb > Zn. The BCF concentration of Zn was (0.349) lower as compared to the present study [36]. The bio concentration depends upon the pH of the soil and they do not easily transfer in the fodder plants [22, 37] Soil pH effects the mobility of metals in soil. Higher the soil pH causes low mobility of metals in soil [38]. It was reported that if the bio concentration factor is less than 1 no accumulation of metals occurred in soil. The higher bio concentration of metal indicates the higher range of its accumulation in plants [39].

Table 9. Bio concentration factor of Pb in fodders having contamination from automobile emissions

Sites	Treatments	<i>P.glaucum</i>	<i>T.alexandrinum</i>	<i>S.officinarum</i>	<i>Z.mays</i>	<i>B.compestris</i>
Site 1	AR	1.1315	1.0272	1.0537	0.6517	0.9160
	NR	0.7659	1.040	1.2910	1.0177	1.2857
Site 2	AR	1.0697	0.9276	0.8579	0.7499	0.8276
	NR	1.0223	0.2259	0.5522	0.5998	0.2767
Site 3	AR	0.2717	0.299	0.2980	1.3746	0.880
	NR	0.7232	0.930	0.823	0.2142	1.1537
Site 4	AR	1.0630	1.0996	5.6463	5.3276	3.0283
	NR	3.1647	4.8887	3.0086	3.4217	0.6096

Table 10. Bio concentration of Zn in fodders from four sites of sampling having heavy traffic load

Sites	Treatments	<i>P.glaucum</i>	<i>T.alexandrinum</i>	<i>S.officinarum</i>	<i>Z.mays</i>	<i>B.compestris</i>
Site 1	AR	1.0558	1.0889	1.0548	0.3628	0.967
	NR	0.9499	0.8757	2.7011	0.898	0.1142
Site 2	AR	1.2595	1.2485	0.4367	0.484	0.8974
	NR	1.086	0.3212	0.5816	0.6317	0.3287
Site 3	AR	0.3291	0.2483	0.2294	1.1511	0.8903
	NR	0.8256	0.8501	0.7389	0.2737	0.7694
Site 4	AR	0.772	0.7855	4.566	3.4699	3.9075
	NR	4.4998	4.8043	4.7406	5.1375	5.0255

3.9. Correlation

The positive and non-significant correlation was found in Soil-*P. glaucum* and Soil-*B. compestris* and the negative non-significant correlation was present in Soil-*T.alexandrinum*, Soil-*S.officinarum* and Soil-*Z.mays* (Table 13). The Zn correlation coefficient showed the negative non-significant correlation in Soil-*S.officinarum*, Soil-*Z.mays* and Soil-*B. compestris* (Table 14). The positive non-significant correlation was present in Soil-*P.glaucum* and Soil-*T.alexandrinum*. The current study showed the non-significant correlation of Pb and Zn. According to Onjefu *et al.*, [53] Fe, Mn, Ni, Zn Cr showed the strong ($r > 0.5$ to 0.9) and extremely strong ($r > 0.9$) positive correlation while the Pb showed weak positive correlation ($r=0.3524$). According to Nazzal *et al.* [54] Pb, Mn, Ni, Fe, Cd have showed the significant positive correlation As compared with current study in which the heavy metal Pb showed negative non-significant correlation.

Table 11. Correlation between Pb concentrations of soil to fodders from four road sites

Correlation Coefficient	
Soil- <i>P.glaucum</i>	0.269
Soil- <i>T.alexandrinum</i>	-0.011
Soil- <i>S.officinarum</i>	-0.411
Soil- <i>Z.mays</i>	-0.476
Soil- <i>B. compestris</i>	0.518

Table 12. Correlation between Zn concentrations of soil to fodders from four road sites

Correlation Coefficient	
Soil- <i>P.glaucum</i>	0.523
Soil- <i>T.alexandrinum</i>	0.298
Soil- <i>S.officinarum</i>	-0.103
Soil- <i>Z.mays</i>	-0.105
Soil- <i>B.compestris</i>	-0.170

3.10. Daily intake and health risk index

The Highest value of Daily intake of Pb was found in nearby road sample of *Z.mays* of site 4 while the lowest value was recorded in *T.alexandrinum* of site 2 (Table 15). The Highest value of Health risk of Pb was also found in the *Z.mays* of site 4 collected near the road while the lowest value was recorded in the *B.compestris* of site 3 collected near the road (Table 16). The highest daily intake concentration of Zn was calculated in site 1 *T.alexandrinum* sample taken far away from the road. The lowest value was found in the away road sample of *S.officinarum* of site 2. The highest concentration of Zn in health risk index was found in the *B.compestris* of site 1 of away road sample while the lowest value was found in the away road sample of *S.officinarum* of site 2. The order of Daily intake of metals in animals was in the order Pb > Zn in the current study. The daily intake of Pb was lower in the present study as compared to given by Guerra *et al.*, [42]. The daily intake of metals in the present study lower than 1, indicates no risk of health for livestock via consumption as reported by Radwan and Salama [43]. Our present study showed the order of health risk index as Pb > As > Cd > Fe > Zn > Ni > Mn > Cr. The health risk index was obtained by using the oral reference dose of the metals. The present study showed the health risk index of Pb, Cr, Cd, Zn were higher as compared to Ali *et al.*, [44]. but the health risk of Ni was lower in the present study. According to Khan *et al.*, the health risk of Mn was higher as compared to present study. The concentration of heavy metal greater than 1 showed high risk of health for livestock Sajjad *et al.*, [45]. The metal accumulation even at very low level in liver and kidney cause abnormality of metabolism and health risk for both wildlife and livestock [46,47]. Forages growing in polluted environment such as along roadsides accumulates toxic metals in high concentration which poses danger to the life of animals and enters the food chain and cause detrimental effects [48]. Increase concentration of lead in the forage samples along roadsides cause restriction of root expansion and cell division and cell elongation [49]. The highest mean concentration of lead from vehicle exhaust enters the food chain and cause serious problems in kidneys, liver, central nervous system, reproductive system and anemia in humans [50].

Table 13. Daily intake of Pb and health risk from fodders of the four road sites

Sites	Treatments	DIM&HRI	<i>P.glaucum</i>	<i>T.alexandrinum</i>	<i>S.officinarum</i>	<i>Z.mays</i>	<i>B.compestris</i>
Site 1	AR	DIM	0.0136	0.0145	0.013	0.0083	0.013
		HRI	3.89	4.168	3.8001	2.3915	3.77
	NR	DIM	0.0120	0.012	0.0146	0.0143	0.009
		HRI	3.431	3.5759	4.186	4.095	2.85
Site 2	AR	DIM	0.00857	0.0101	0.0070	0.0063	0.0092
		HRI	2.4496	2.897	2.00384	1.824	2.635
	NR	DIM	0.0094	0.004	0.008	0.0094	0.00427
		HRI	2.7103	1.228	2.5709	2.707	1.222
Site 3	AR	DIM	0.0044	0.0058	0.00595	0.0227	0.017
		HRI	1.2643	1.684	1.7025	6.513	4.87
	NR	DIM	0.0184	0.018	0.0183	0.0048	0.025
		HRI	5.2789	5.386	5.2452	1.374	7.19



Site 4	AR	DIM	0.022	0.025	0.0222	0.020	0.020
		HRI	6.4907	7.229	6.37	5.89	5.962
	NR	DIM	0.0230	0.019	0.0208	0.023	0.023
		HRI	6.599	5.62	5.944	6.695	6.63

Table 14. Daily intake and health risk of Zn from fodders polluted with traffic exhaust from four sites

Sites	Treatments	DIM&HRI	<i>P.glaucum</i>	<i>T.alexandrinum</i>	<i>S.officinarum</i>	<i>Z.mays</i>	<i>B.compestris</i>
Site 1	AR	DIM	0.0203	0.0208	0.0205	0.008	0.021
		HRI	0.067	0.069	0.0686	0.027	0.0712
	NR	DIM	0.0206	0.0197	0.0208	0.019	0.0090
		HRI	0.068	0.065	0.069	0.066	0.03
Site 2	AR	DIM	0.0081	0.00911	0.0029	0.004	0.0069
		HRI	0.027	0.030	0.0099	0.013	0.023
	NR	DIM	0.0097	0.0044	0.0074	0.008	0.0036
		HRI	0.0326	0.014	0.0249	0.0292	0.012
Site 3	AR	DIM	0.0038	0.0033	0.0037	0.0156	0.0120
		HRI	0.0129	0.0110	0.0126	0.052	0.040
	NR	DIM	0.0142	0.0126	0.0128	0.0045	0.011
		HRI	0.0475	0.0421	0.0426	0.015	0.039
Site 4	AR	DIM	0.0130	0.0121	0.0169	0.015	0.016
		HRI	0.043	0.040	0.0565	0.0507	0.055
	NR	DIM	0.016	0.0174	0.0158	0.0170	0.0169
		HRI	0.0551	0.058	0.0528	0.056	0.056

3.11. Pollution Load index

The highest pollution load index of Pb was examined in *P.glaucum* growing near road of site 3 and lowest was found in away road sample of *Z.mays* of site 4 (Table 17). The pollution load index of Zn in fodders soil was found highest in *T.alexandrinum* of site 1 collected near the road while the lowest concentration was found in the *Z.mays* of site 4 taken near the road side (Table 18). The pollution load index of Zn, Pb, Mn and Fe was higher as compared to the present study. The pollution load index of Pb and Zn in the current study is lower as obtained by Mmolawa [35]. The pollution load index will be equal to 1 than plants are contaminated by pollutants, if $PLI < 1$ it shows excellence level and if $PLI > 1$ it shows the site is deteriorated with anthropogenic activities [40]. As in the present study the highest concentration of pollution load index Of Zn showed that it directly relates with traffic density and distance from the road. As the distance of soil sample from the road decreases the level of its concentration becomes low [41].

Table 15. Pollution Load Index of Pb in soil samples from four road sites

Sites	Treatments	<i>P.glaucum</i>	<i>T.alexandrinum</i>	<i>S.officinarum</i>	<i>Z.mays</i>	<i>B.compestris</i>
Site 1	AR	0.07336	0.0864	0.0768	0.0782	0.0878
	NR	0.0954	0.0732	0.0691	0.0857	0.0473
Site 2	AR	0.0488	0.0665	0.0497	0.0518	0.0678
	NR	0.0565	0.1159	0.0992	0.0962	0.0941
Site 3	AR	0.0991	0.1198	0.1217	0.101	0.1180
	NR	0.1555	0.1233	0.1358	0.1366	0.1328
Site 4	AR	0.13011	0.1401	0.024	0.0235	0.0419
	NR	0.0444	0.0245	0.0421	0.0417	0.231

Table 16. Pollution Load Index of Zn in soil from four sampling sites facing high traffic density

Sites	Treatments	<i>P.glaucum</i>	<i>T.alexandrinum</i>	<i>S.officinarum</i>	<i>Z.mays</i>	<i>B.compestris</i>
Site 1	AR	0.200	0.1978	0.2021	0.2315	0.2285
	NR	0.224	0.2339	0.0799	0.228	0.8175
Site 2	AR	0.066	0.0755	0.0711	0.0872	0.0802
	NR	0.0932	0.1431	0.1333	0.144	0.1146
Site 3	AR	0.1223	0.1379	0.1710	0.1411	0.1401
	NR	0.1786	0.1538	0.17936	0.1713	0.1580
Site 4	AR	0.1750	0.1601	0.0384	0.045	0.0438
	NR	0.038	0.0376	0.034	0.034	0.0348

3.12. Enrichment factor

The highest value of Pb concentration was observed in *S.officinarum* of site 4 collected away from the road, while the lowest value was observed in *Z.mays* of site 3 collected from the roadside. The highest enrichment factor of Zn metal was determined in near road sample of *Z.mays* while the lowest concentration was present in away road sample of *B. compestris* of site 2. Sulaiman and Hamzah, reported the enrichment factor of Pb and Cd which was lower as compared to the current study [51]. The enrichment factor of Pb in the present study was higher. Their study showed the range of enrichment factor from moderate contamination to considerable contamination. Enrichment factor is the concentration of heavy metal present in the soil. The present study showed the order of enrichment factor as Zn>Pb. The study reported by Amieret *al.*, the concentration of Pb showed extremely high enrichment as compared to present study which showed the lower enrichment concentration [52].

Table 17. Enrichment factor of Pb in soil polluted with automobile emissions from four sites of sampling

Sites	Treatments	<i>P.glaucum</i>	<i>T.alexandrinum</i>	<i>S.officinarum</i>	<i>Z.mays</i>	<i>B.compestris</i>
Site 1	AR	48.09072	43.65869	44.78521	27.698	38.93402
	NR	32.55534	44.20233	54.86942	43.25444	54.64728
Site 2	AR	45.46648	39.42603	36.46178	31.87387	35.1777
	NR	43.45199	9.604651	23.47072	25.49359	11.76228
Site 3	AR	11.54785	12.73879	12.66772	58.4228	37.43114
	NR	30.73996	39.5567	34.98788	9.106884	49.03556
Site 4	AR	45.18216	46.73487	239.9747	226.4312	128.7067
	NR	134.5056	207.7746	127.8679	145.4257	25.91066

Table 18. Enrichment factor of Zn in soil polluted with vehicular emissions from four sites of sampling

Sites	Treatments	<i>P.glaucum</i>	<i>T.alexandrinum</i>	<i>S.officinarum</i>	<i>Z.mays</i>	<i>B.compestris</i>
Site 1	AR	87.99069	90.74857	87.90598	30.23355	80.64916
	NR	79.16154	72.98211	225.0967	74.8925	9.518858
Site 2	AR	104.9585	104.0471	36.39709	40.36079	74.78979
	NR	90.55664	26.77149	48.4713	52.64236	27.39258
Site 3	AR	27.42892	20.69563	19.12294	95.93042	74.19448
	NR	68.80505	70.84341	61.57727	22.81615	64.12195
Site 4	AR	64.39506	65.4643	380.56	289.1617	325.6327
	NR	374.9912	400.3614	395.0573	428.1305	418.792

4. Conclusions

Atmospheric pollution is the major cause of these metals in soil and plants. Vehicle exhaust is the source of atmospheric pollution. The current study reported that the natural weathering of building materials, emissions from industries, bricks kilns, and heating systems burning of fuel also cause the heavy metal pollution but the vehicles are also the source of its accumulation in plant and animal bodies.

The pollution of heavy metals from automobiles is of serious environmental issue. Due to increase in population the traffic burden is also increasing day by day. The present study was done on the fodder samples along roadside environment. Fodder plants and soil are facing high concentration of heavy metal

pollution originating from busy roads in the vicinity of suburbs and contribution of high traffic density to high level of exhaust emissions.

Samples analyzed in the current study have low and high values of heavy metals from the permissible limits. The heavy metal concentration is strongly and inversely correlated with distance from the road. Lead and Zinc are major metals present in roadside environment and they are released from burning of fuels, tyres and through leakage of oils. Heavy metal from vehicles exhaust are very toxic for living organisms. Even some trace elements are essential for plants but they become toxic if their occurrence exceed the limit.

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