Synthesis and Characterization of 1-Ethylamide and 1-Ethanolamide of D-Cloprostenol and their 15-Epimers

CONSTANTIN TANASE1*, DENISA IOANA UDEANU2*, CONSTANTIN DRAGHICI3, FLOREA COCU4

1National Institute for Chemical -Pharmaceutical Research and Development ICCF, 112 Vitan Av., 031299, Bucharest, Romania
2University of Medicine and Pharmacy Carol Davila, Bucharest, Faculty of Pharmacy, Department of Clinical Laboratory and Food Safety, 6 Traian Vuia, 20956, Bucharest, Romania
3Organic Chemistry Center C.D.Nenitescu, 202B Splaiul Independentei, 71141, Bucharest, Romania

1-Ethylamide-, 1-ethanolamide of D-Cloprostenol and their 15-epi-isomers were synthesized by amidation of methyl esters of D-Cloprostenol or 15-epi-Cloprostenol. The crude compounds were purified by pressure chromatography on silica gel and fully characterized by IR, 1H-, 13C-, 2D-NMR (COSY and HETCOR) and MS spectroscopy.

Keywords: D-Cloprostenol-1-ethylamide, D-Cloprostenol-1-ethanolamide, 15-epi impurities, prostamide, amides, synthesis of amides

Since the discovery of prostaglandins [1] and the characterization and chemical classification of the multitude of structural prostaglandin compounds discovered and synthesized [2], a lot of modifications of the 1-carboxyl group were performed with the goal to reduce the side effects of the acid group and to increase the penetration of the cell wall. The intention was to replace the carboxy group by a non-ionizable group. The first step was to replace the carboxy with an ester. At the beginning, the methyl esters were easily synthesized and the corresponding compounds were investigated [3], but then the isopropyl ester was found later to be the ester most used in the prostaglandin active substances (PGF, isopropylester was considered in 1987-1989 to be the most potent ocular hypotensive agent ever reported [4, 5]), like in (+)- or (±)-Cloprostenol isopropilester, Unoproston isopropylerster, Latanoprost, Fluprostenol, Travoprost, Tafuprost, which are mainly used as drugs in the reduction of intraocular pressure. Other transformations were also performed; the reduction of the carboxy to a primary alcohol [3,6], the replacement of the carboxy group with a primary or secondary amine [7,8] and the transformation of the carboxy group with an amide [7-9].

In the amide group there is an increased resonance between the nitrogen and carbonyl which makes the molecule have some specific characteristic properties. Firstly, the hydrolysis of the amide group is at least two orders of magnitude slower than the hydrolysis of the esters and its hydrolysis in vivo is somewhat similar to what happens in the prodrugs, but not at the same extent. Secondly, the increased resonance between the nitrogen and the carbonyl modifies the C-N sigma bond in such a way that the rotation barrier is greater than in the C-O group of esters or carboxylic acids. The result is that the prostamide molecule becomes more rigid, more stable in a sterical conformer, which has a significant effect for binding to a number of receptor sites in the pocket geometry. Also, the hydrogen bonding to the receptor sites is changed. All of these changed characteristics confer the prostamides significantly different biological properties than those of prostaglandins.

The progression from the natural prostaglandin amides to the prostaglandin amide analogues was mainly accelerated by two factors:

1. The reduction of the intraocular pressure of the Bimatoprost, a 1-ethylamide analogue of 17-phenyl PGF2α (fig. 1) [4, 9-11]; this compound has the most efficacious ocular hypotensive activity, greater than that of timolol and Latanoprost [11]. A side effect of the prostaglandins used in reducing the intraocular pressure was efficiently exploited by the use of Bimatoprost not only as a drug, but also as cosmetics for increasing the length, thickness and darkness of eyelashes in patients with hypotrichosis [9, 11-13], and its use in cosmetic products surpassed its use in ocular treatment; this effect was also found and exploited for other isopropyl esters of prostaglandin analogues used for the reduction of ocular pressure, like Latanoprost [13].

2. The discovery of the biological activity of anandamide (ethanolamide of arachidonic acid) as a substrate for cyclooxygenase-2[14] and generally the elucidation of the biosynthetic pathway from anandamide to the big family of prostamides (fig. 1).

Numerous chemical structures were synthesized [9] and the number of claimed structures is impressive. The research in prostamides is on a raising trend and has extended also to other prostaglandin types, like PGE[15].

In the patent literature, amides on a Cloprostenol structure were claimed by someone, but neither the explicit synthesis nor the physico-chemical characterization of amides with an ethyl or ethanol substituent were published. For many years, we have used some amides from

* email: cvtanase@gmail.com; denisaudeanu@gmail.com

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Fig. 1. Significant prostamides which accelerated the synthesis of new prostamide analogues
2. Synthesis of epi-D-Cloprostenol ethylamide, 5

The compound was obtained by the same procedure from a fraction of epi-D-Cloprostenol Methyl ester containing ~30% D-Cloprostenol-methyl ester. (silica gel, ethyl acetate-methanol-acid acetic, I, 90:13:1, Rf = 0.49). The product was purified by multiple pressure chromatography on silica gel (eluent: ethyl acetate-methanol, 90:13), resulting a pure fraction of epi-D-Cloprostenol 1-ethylamide as oil, IR: 3306br vs, 2930s, 2873m, 1642s, 1593s, 1575s, 1575s, 1479s, 1454m, 1428m, 1284m, 1243m, 1210m, 1031s, \(\text{δ} = 75.00 \text{MHz} \) for 1H and 75 MHz for 13C) or Bruker 400 MHz (400 MHz for \(\text{δ} = 100 \) and 100 MHz for \(\text{δ} = 13C\)); the chemical shifts are given in ppm relative to TMS as internal standard.

Complementary spectra: COSY, HETCOR and with trifluoroacetic acid added were done for the correct assignment of the NMR signals. The numbering of the atoms in the compounds is presented in schemes. The purity of the prostamides used in the testing experiments was established by HPLC on a LA CHROM ELITE HITACHI with diode array (DAD) detector on a Kromasil column (250 x 4.6 mm, C-18 silicagel, 5 mm), mobile phase: methanol-water-glacial acetic acid, 55:44:3.07 (v/v), 1.0 mL/min., 35°C, \(\text{λ} = 257 \text{nm} \).

3. Synthesis of D-Cloprostenol 1-ethanolamide, 3

4.39 g (10 mmoles) D-Cloprostenol methyl ester were dissolved in 25 mL methanol, 10 mL ethanolamine were added, then 16 mL 25% MeOna in MeOH (70 mM) solution were added in an argon atmosphere. The mixture was stirred at room temperature for two days, while monitoring the reaction by TLC (silica gel, I, ethyl acetate-methanol acid acetic, I, 90:13:1, Rf = 0.63, Rf = 0.25). The reaction mixture was cooled on an ice-water bath, the base was neutralized with 13 mL acetic acid and concentrated under reduced pressure. The residue was taken in 70 mL water and 150 mL ethyl acetate, the organic phase was separated (aqueous phase was extracted further with 5x 75 mL ethyl acetate), washed with 70 mL sat. soln. NaHCO\(_3\), 75 mL brine, dried (Na \(_2\)SO\(_4\)), filtered and concentrated. The crude product was purified by pressure chromatography on silica gel (eluent: dichloromethane-glacial acetic, I, 90:13:1, \(\text{Rf} = 0.66, \text{Rf} = 0.25 \)). The purified product was obtained through silica gel chromatography on silica gel column eluted with dichloromethane, then with the solvent system: dichloromethane-isopropanol in ratios from 98:2 to 95:5. Pure D-Cloprostenol 1-ethyl amide \(\text{δ} = 0.63\) was obtained as an colorless oil, [\(\alpha\)] \(_D\) \(= +27.9^\circ\) (1% in ethanol), IR: 3298br vs, 2930s, 1631s, 1596s, 1579s, 1553s, 1480s, 1454m, 1428m, 1284m, 1243m, 1210m, 1031s, \(\text{δ} = 75.00 \text{MHz} \) for 1H and 75 MHz for 13C) or Bruker 400 MHz (400 MHz for \(\text{δ} = 100 \) and 100 MHz for \(\text{δ} = 13C\)); the chemical shifts are given in ppm relative to TMS as internal standard.

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H2O, 6.0, with TFA), CH2OH, 54.81 (C-12), 49.48 (C-8), 43.95 (C-10), 41.42 (CHNH), 34.93 (C-2), 26.34 (C-7), 25.36 (C-4), 24.80 (C-3). C18H24NO2. M. wt. 467.982. MS th. [M+H]+: 468.21474, found: [M+H]+: 468.21391;

4. Synthesis of epi-D-Cloprostenol ethanolamide, 6

The synthesis of epi-D-Cloprostenol ethanolamide, 6, was realized in the same way as presented above for D-Cloprostenol ethanolamide 3: from 440 mg (1 mmole) epi-D-Cloprostenol methyl ester, slightly impurified with D-Cloprostenol methyl ester, 320 mg (68.4%) epi-D-Cloprostenol ethanolamide were obtained as an oil, used in HPLC to identify this impurity in compound 3; TLC (silica gel, ethyl acetate-methanol-acetic acid, 90:13:1, Rf = 0.66, Rf = 0.29), [α]D = +17.8° (1% in ethanol); IR: 3308 br vs, 2930s, 2875m, 1638ms, 1594s, 1579m, 1552, 1516s, 1438m, 1300s, 974w, 908w, 770w, 1H-NMR(DMSO-d6, δ ppm, J Hz): 7.29 (t, 1H, H-5', 8.2), 7.00 (t,1H, H-2', 2.1), 6.97 (dd, 1H, H-4', 0.9, 1.9, 8.2), 6.97 (dd, 1H, H-5', 7.7, 15.6), 5.52 (dd, 1H, H-11, 5.0, 15.6), 5.48 (m, 1H, H-5 or 6), 5.27 (dt, 1H, H-6 or 5, 7.1, 11.0), 5.15 (d, 1H, 15- OH, 4.9, deuterable + TFA), 4.65 (m, 1H, CH,OH), 4.55 (d, 1H, 9-OH, 6.0, deuterable + TFA), 4.38 (d, 1H, 11-OH, 5.0, deuterable + TFA), 4.30 (m, 1H, H-15), 3.94 (dd, 1H, H-1, 9.8, 11.5), 3.83 (dd, 1H, H-16, 7.1, 15.6), 3.68 (m, 1H, H-11; dt with TFA, 7.1, 15.6), 3.09 (q, 2H, CH2NH, 7.2, 7.7), 2.24-2.13 (m, 1H, H-10, H-12), 2.05 (m [two overlapped triplets], 4H, 2H2, 2H-3, 7.2, 7.7), 1.97 (br q, 2H, H-7, 7.2), with TFA [1.62, t, 1H, H-4, 7.4, with TFA], 1.48 (t, 1H, H-4.7, 7.3, with TFA) 1.44 (dddt, 1H, H-10, HETCOR 2.2, 5.5, 14.0), 1.33 (ddtt, ddt with TFA, 1H, H-8, 5.0, 6.3, 10.0). 13C-NMR(DMSO-d6, δ ppm): 173.82 (C-1), 160.28 (C-1'), 154.31 (C-3), 130.15 (C-13), 131.44, 131.34 (2C, C-5, C-6), 130.13 (C-5'), 129.54 (C-14), 121.05 (C-4'), 115.26 (C-2'), 112.47 (C-6'), 76.25 (C-9), 73.19 (C-16), 70.13 (C-15), 69.78 (C-11), 65.07 (CH,OH), 54.81 (C-12), 49.48 (C-8), 44.57 (C-10), 42.03 (CHNH), 35.54 (C-2), 26.97 (C-7), 26.00 (C-4), 25.37 (C-3). C18H24NO2. M. wt. 467.982. MS th. [M+H]+: 468.21474, found: [M+H]+: 468.21391; [450 (C18H24NO2), 322 (C18H26NO2), 304 (C18H28NO3), 286 (C18H30NO4)].

The purities of the prostamides were established by HPLC and the retention times were: 26.19 for compound 2, 15.99 for compound 3 and 13.67 for compound 6. No corresponding epi-5 was detected in HPLC for compound 2 (the purity of the compound was 98.7 %) and the compound 3 contained 1.6 % epi-impurity 5 and had a 98.2 % purity; these compounds were used in their conditioning for ocular treatment applications.

Results and discussions

The synthesis of amides has been extensively studied for obtaining new oleamide analogues [18]. C-1 unsubstituted prostamides were obtained by treating the corresponding prostaglandin methyl ester with ammonia and NHCl in a sealed tube [8] and N-alkylamides were obtained by the reaction with primary amines [7], even with aqueous amines like 75% aqueous EtNH2, [19]. Other methods give the amidation of free acid with amines by activation acid with carbonyldimidazole [15], DCC [20] or chloroformamide [21]. Though the ethylamide or ethanol-amide of Cloprostenol are possibly claimed in some patent, neither the synthesis of these compounds nor their full characterization was presented. For a long time, we have synthesized both of the compounds mentioned above, and these compounds were used in some biological evaluations [15,16]. The extensive biological experiments of D-Cloprostenol 1-ethylamide 2 and 1-ethanolamide 3 (Scheme 1) in reducing the intraocular pressure and in dermal treatment like in the case of Bimatoprost and more recently of Latanoprost, require increased quantities of not only of the compounds described above, but also of their 15-epi-counterparts, 5 and 6 (Scheme 2). 15-epi-D-Cloprostenol 1-ethylamide 5 and 1-ethanolamide 6, are inactive by-products in the active substances 2 and 3, originated from the total stereoselective synthesis of Cloprostenol at the stage of the stereoselective reduction of the corresponding enone to the allylic alcohol. By the manufacturing regulations, the level of 15-epi impurities was continuously decreased, now being near 1% in the active compound.

For obtaining the 1-ethylamide compounds, 2 and its 15-epimer 5, we have chosen as method the amidation of an ester with an amine. As ester we used D-Cloprostenol methyl ester for obtaining 2, and an impure fraction of 15-epi-D-Cloprostenol methyl ester (containing near 30% D-Cloprostenol methyl ester) for obtaining 5. The amidation was performed in a non-aqueous solvent, with ethylamine in methanol ((50°C, 72 h), not with 70% aq. EtNH2, as mentioned in the literature [4a, 4b], compound 2 being obtained in ~86% yield. A stream of EtNH2 was introduced in a methanol solution of the cyclic ester of α-side chain (prepared with the 9-hydroxy group (protected at the hydroxyl groups as 11,15-bis-THP ether) for 3-4 h at 10°C, followed by 65-70 h at 25-30°C (no yield data), was also mentioned in a patent for obtaining bimatoprost [4c]. The amidation of the above unprotected cyclic ester of α-side chain with the 9-hydroxy group was performed with 2M EtNH2 in THF (40°C, 18 h, 73.8% for obtaining bimatoprost [4d]. The amidation of D-Cloprostenol isoproxyester for obtaining 2 was a very slow reaction, even at elevated temperature.

The amidation of D-Cloprostenol methyl ester 1 with ethanolamine was done in methanol in the presence of sodium methoxide in excess (7:1) as base, at r.t., in over 79% yield. In the literature, the amidation is presented with a slower excess (1:1:1) of sodium methoxide in non-alcoholic solvent, at reflux with removing the alcohol formed in the reaction [21] and also in a catalyst amount (5%MeONa) in almost the same conditions [22]. We have chosen to develop the reaction with an excess of MeONa in methanol at r.t., the conditions being suitable for the structure of Cloprostenol, which has a chlorine atom linked in meta on the aromatic ring of the molecule.

After the usual work-up, the compounds were purified by pressure chromatography and the pure prostamide analogues were obtained as a colorless or slightly yellow oil, then they were characterized by [α]D, IR, MS, 1H- and 13C-NMR analysis (Materials and Methods for details). Epi-D-Cloprostenol ethanolamide 5 and epi-D-Cloprostenol ethanolamide 6 were synthesized from the epi-D-cloprostenol methyl ester 4 by the same methods presented for compounds 2 and 3 (Scheme 2). Compounds 5 and 6 were synthesized to be tested for their activity by comparison with that of 2 and 3, and also to be used as...
of the pure compounds 2 and 3, in the active substance and also in the conditioned pharmaceutical drugs.

The purity of the D-Cloprostenol 1-ethyl and 1-ethanol amides was established by HPLC, the compound 2 being obtained in 98.7% and the compound 3 in 98.2% purity.

Conclusions

1-Ethyl and 1-ethanolamides of D-Cloprostenol, 2 and 3, were obtained by the amidation of D-Cloprostenol methyl ester with ethylamide in methanol, respectively with ethanolamide in the presence of sodium methoxide as catalyst. The corresponding 15-epimers, 5 and 6, were obtained from an impure fraction of epi-D-Cloprostenol methyl ester (containing ~30% D-cloprostenol methyl ester) by the same procedures as for 2 and 3. The compounds were purified by pressure chromatography, their purity was determined by HPLC and then they were fully characterized by optical rotation, IR, 1H-, 13C-, 2D-NMR, and high resolution MS spectroscopy.

References