

Investigation of linearity, detection limit (LD) and quantitation limit(LQ) of active substance from pharmaceutical tablets

AFRODITA DOINA MARCULESCU¹, CRISTIAN-CATALIN GAVAT^{2*}, AUREL NECHITA^{3*}, GABI TOPOR³,
LEONARD VASILE VASILESCU¹, MIHAELA DEBITA³, ELENA ROXANA AXENTE³, LUCIA CARMEN TRINCA⁴, LUCRETIA ANGHEL³

¹ Grigore T. Popa University of Medicine and Pharmacy, Faculty of General Medicine, Department of Biochemistry, 16 Universitatii Str., 700115, Iasi, Romania.

² Grigore T. Popa University of Medicine and Pharmacy, Faculty of Medical Bioengineering, Department of Biomedical Sciences, 16 Universitatii Str., 700115, Iasi, Romania

³ Dunarea de Jos University of Galati, Medicine and Pharmacy Faculty, Department of Dentistry, 47 Domneasca Str., 800008, Galati, Romania

⁴ Ion Ionescu de la Brad University of Agricultural Sciences and Veterinary Medicine of Iasi, 3 M. Sadoveanu Alley, 700490, Iasi, Romania

The aim of this research was to exactly quantify pure sodium metamizole from tablets, using a spectrophotometric analysis in Visible range. The method applied has been subjected to a validation protocol which consisted in analyzing the following parameters: linearity of the method, detection limit (LD), quantitation limit (LQ).

Following actual dosing, pure sodium metamizole amount in tablet of pharmaceutical was found to be 477.477 mg assigned to a percentage content of 95.495%, very close to official declared amount (500 mg), with an maximum average percentage deviation of only 4.505% from the official declared active substance content. This value was situated below the maximum admissible percentage deviation from stated active substance content ($\pm 5\%$), established by Romanian Pharmacopoeia, X-th Edition rules.

Keywords: sodium metamizole, detection limit, quantitation limit

Metamizole (dipyrone, as sodium salt) is a popular analgesic medicine, non-opioid drug, commonly used in human and veterinary medicine similar to other natural or synthetic active principles such as black pepper, lavender, pepper, curcumin ibuprofen [1-6]. In some cases, this agent is still incorrectly classified as a non-steroidal anti-inflammatory drug (NSAID). Apart from its strong analgesic effect, the medication is an moderate antipyretic and significant spasmolytic agent [7-12]

The spasmolytic effect of metamizole is a result of mechanism associated with a powerful inhibition of intracellular calcium (Ca^{2+}) release, as a result of the reduced inositol phosphate synthesis. Metamizole is predominantly applied in the therapy of pain of different etiology, of spastic conditions, especially affecting the digestive tract[13], and of fever refractory to other treatments. It is especially indicated as an strong, effective analgesic in all types of moderate and intense pain (neuralgia, arthralgia, myalgia, headache, dysmenorrhoea), including postoperative pain, renal and biliary colic, dental pain[14-17]

Experimental part

Method and procedures

Algoalmin(sodium metamizole) was oxidized by 5.0% ammonium orthomolybdate ($(\text{NH}_4)_2\text{MoO}_4$ aqueous solution in a strongly acidic medium (H_2SO_4 , 40%), to form a bluish-colored green compound that showed a maximum absorption to $\lambda = 690 \text{ nm}$ (fig 1).

Visible absorption spectra of green-bluish compound synthesized. Evaluation of maximum absorption wavelength, specific absorptivity and molar absorption coefficient

Sample synthesis from Algoalmin® Zentiva tablets

One Algoalmin tablet was weighted to find the average mass of pharmaceutical product it was found that mean mass value was $m_c = 0.5333 \text{ g}$. The official declared content by pharmaceutical company of pure sodium metamizole in tablet was 500 mg. Then, 3 tablets were finely crushed and a = 0.1102 g of Algoalmin obtained powder were quantitatively brought with a little volume of absolute methanol (8 mL) into a $V = 100 \text{ mL}$ volumetric flask. The content was mixed until complete dissolution of sodium metamizole and filled up to the mark with distilled water.

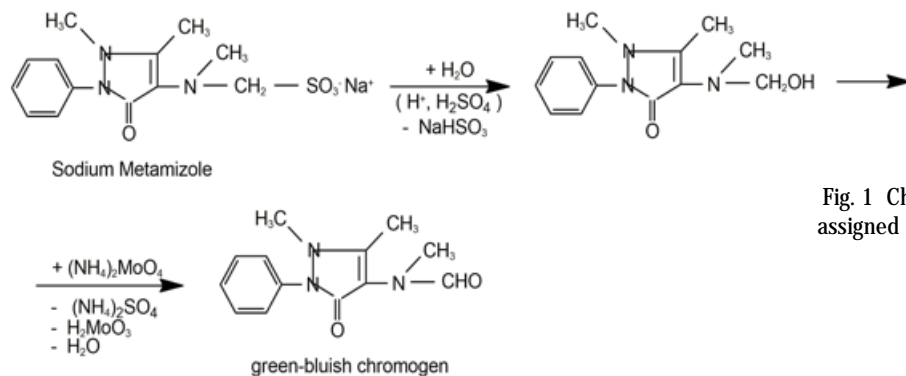


Fig. 1 Chemical reactions of Sodium Metamizole assigned of green-bluish chromogen synthesis

* email: ccgavat70@yahoo.com Phone: 0743-782544, nechitaurel@yahoo.com

From the obtained sample solution, $v_1 = 0.4$ mL were measured and quantitatively brought to 10 mL graduated glass tube. Then, 1.5 mL of ammonium orthomolybdate $(\text{NH}_4)_2\text{MoO}_4$, 5.0 % and 0.5 mL H_2SO_4 , 40% were added. (table 3). Sample solution was stirred well, stored in a dark place for 30 minutes and filled up to volume $V_p = 10$ mL with distilled water.

Five measurements have been made and sample mean absorbance A_p , was calculated. (T. 3).

Calculation method

Calculation of pure sodium metamizole amount in pharmaceutical tablet

The average measured mass of one tablet of pharmaceutical product was $m_c = 0.5333$ g (533,3 mg). According to manufacturing company, a pharmaceutical tablet of Algocalmin[®] contained 500 mg of pure sodium metamizole. The amount of pure sodium metamizole existing in final volume of sample solution (V_p) was determined as follows: $X = C_s (\mu\text{g/mL}) \cdot V_p$ (2), whereas $V_p = 10$ mL has represented sample solution final volume contained in graduated glass tube.

The quantity of pure sodium metamizole from $V_1 = 100$ mL (volumetric flask) was calculated: $X_1 = (V_1 \cdot X) / v_1$ (3), $v_1 = 0.4$ mL was sample solution volume measured from volumetric flask and quantitatively brought to $V_p = 10$ mL graduated test tube.

The amount of pure sodium metamizole in tablet of Algocalmin[®] was investigated as follows: $Y_1 = (m_c \cdot X_1) / a$ (4), whereas $a = 0.1102$ g fine powder sample of Algocalmin[®], prepared from pharmaceutical tablets. Y_1 has represented the amount of pure sodium metamizole in tablet sample, expressed as μg pure sodium metamizole / tablet, then by transforming into mg pure sodium metamizole / tablet of Algocalmin[®] [18-25]

Table 1

MAXIMUM PERCENTAGE DEVIATIONS FROM THE STATED CONTENT OF ACTIVE SUBSTANCE IN PHARMACEUTICALS

Declared content of active substance	Maximum accepted percentage deviations
up to 10 mg	± 10 %
10 mg and up to 100 mg	$\pm 7,5$ %
100 mg and over 100 mg	± 5 %

Percentage content of pure sodium metamizole in commercial tablet (Z %).if it was known that one tablet of Algocalmin[®] had 500 mg of pure sodium metamizole:

$Z = (100 \cdot Y_1) / 500$ (%), so $Z = Y_1 / 5$ (%) (5), Y_1 has been expressed in mg .

Linearity of the method. Regression line parameters. The linearity of analysis process consisted of the ability to lead to results directly proportional to the concentration of an analyte in a given sample, within a given range (1-40 μg /mL), Practically, the intensity of analytical signal (measured absorbance) has varied in directly proportion to the concentration, for a given area. Correlation coefficient had to be $R > 0.999$ and linear regression coefficient $R^2 \approx 0.999$ [10, 11]. The statistic parameters of method linearity were then determined, using Microsoft Office Excel 2016 software and described in table 4.

Detection limit (LD) was the smallest amount of analyte that could be detected in a sample compared to a blank, under established experimental conditions. It was expressed in the same units as concentration of the analyte ($\mu\text{g/mL}$) and was evaluated using formula: $LD = 3 \cdot SE / \text{slope}$ (6). SE has represented standard error of the regression line [18-26]

Quantitation limit (LQ) was given by the lowest analyte concentration in a sample, which could be quantified (determined) with acceptable precision and accuracy under the same experimental conditions. Its value was expressed in the same units as analyte concentration ($\mu\text{g/mL}$) and was calculated as follows: $LQ = 10 \cdot SE / \text{slope}$ (7) [18-27].

Results and discussions

Visible absorption spectra of green-bluish compound. Maximum absorption wavelength, specific absorptivity and molar absorption coefficient investigation

Absorption spectra of green-bluish chromogen was plotted for 7 $\mu\text{g/mL}$ solution (fig. 2) and maximum absorption wavelength was determined to $\lambda = 690$ nm. Mean measured absorbance value to this wavelength was $A = 0.1653$. Concentration of 7 $\mu\text{g/mL}$ was transformed as follows: 7 $\mu\text{g/mL} = 0.0007$ g/100 mL and was assigned to a $2.853 \cdot 10^{-5}$ mole/L concentration, with the respect of calculated molecular weight corresponded to green-bluish chromogen, which was $M = 245.338$ g/mole (fig.3).

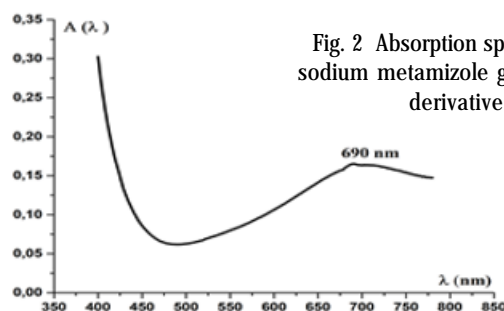


Fig. 2 Absorption spectra (VIS) of sodium metamizole green-bluish derivative

Specific absorptivity:

According to Lambert-Beer's law: $A_{1\text{cm}}^{1\%} = A / C$ (g / 100 mL) (1), $C = 0.0007$ g / 100 mL = solution concentration
 $A =$ calculated mean absorbance = 0.1653
 $A_{1\text{cm}}^{1\%} =$ specific absorptivity

By replacing these values in relation (1), it was obtained: $A_{1\text{cm}}^{1\%} = 0.1653 / 0.0007 = 236.143$. Thus, $A_{1\text{cm}}^{1\%} = 236.143$

Molar absorption coefficient: Similarly, $E_{\text{molar}} = A / C$ (mole / L) (2), $C = 2.853 \cdot 10^{-5}$ mole / L = solution concentration
 $A =$ calculated mean absorbance = 0.1653
 $E_{\text{molar}} =$ molar absorption coefficient

By replacing these values in relation (2), it was obtained: $E_{\text{molar}} = 0.1653 \cdot 10^5 / 2.853 = 5793.901$. Thus, $E_{\text{molar}} = 5793.901$

$M_{\text{green-bluish compound}} = 245.338$ g/mole

Fig. 3 Specific absorptivity and molar absorption coefficient calculation

Table 2
MEASURED ABSORBANCES OF SODIUM METAMIZOLE STANDARD SOLUTIONS

No. det	mL sodium metamizole 100 µg/mL (standard)	mL (NH ₄) ₂ MoO ₄ 5%	mL H ₂ SO ₄ , 40%	C (µg/mL)	A(λ)
1.	0.1	1.5	0.5	1.0	0.0130
2.	0.2	1.5	0.5	2.0	0.0432
3.	0.3	1.5	0.5	3.0	0.0708
4.	0.4	1.5	0.5	4.0	0.0920
5.	0.5	1.5	0.5	5.0	0.1180
6.	1.0	1.5	0.5	10.0	0.2360
7.	1.5	1.5	0.5	15.0	0.3440
8.	2.0	1.5	0.5	20.0	0.4620
9.	2.5	1.5	0.5	25.0	0.5800
10.	3.0	1.5	0.5	30.0	0.7080
11.	3.5	1.5	0.5	35.0	0.8120
12.	4.0	1.5	0.5	40.0	0.9350

Table 3
SAMPLE CONCENTRATION AND SODIUM METAMIZOLE PURE AMOUNTS DETERMINED IN TABLET

Algocalmin® sample	A _s	C _s (µg/mL)	µg sodium metamizole /tablet	mg sodium metamizole /tablet
	0.9204	39.466	477.477	477.477

Linearity of the method -regression line drawing and characteristics

Determined absorbances values of standard solutions measured to $\lambda = 690$ nm, were listed in table 2.

Investigation of sodium metamizole concentration (µg/mL) in Algocalmin® sample solution

Mean absorbance of Algocalmin® sample containing sodium metamizole as active substance, sodium metamizole concentration which was expressed in µg/mL, as well as the amount of pure sodium metamizole calculated in tablet of pharmaceutical product, were shown in table 3.

From relation (1): $C_s = (A_s + 0.0031) / 0.0234$ (µg/mL) = $(0.9204 + 0.0031) / 0.0234$, so $C_s = 39.466$ µg/mL has represented sample concentration of pure sodium metamizole.

Calculation of pure sodium metamizole amount on pharmaceutical tablet Algocalmin®

Pure sodium metamizole amount existing in final volume of sample solution ($V_p = 10$ mL) according to equation (2), was: $X = 39.466 \cdot 10 = 394,66$ µg.

Sodium metamizole quantity from $V_1 = 100$ mL (volumetric flask) was calculated, as follows: $X_1 = (100 \cdot 394.66) / 0.4$, according to relation (3). So, $X_1 = 98665$ µg.

The amount of pure sodium metamizole in tablet of pharmaceutical product: $Y_1 = (0.5333 \cdot 98665) / 0.1102 = 477477.717$ µg sodium metamizole. Thus, $Y_1 = 477,477$ mg pure sodium metamizole / tablet of pharmaceutical Algocalmin®.

Percentage content of pure sodium metamizole in commercial tablet (Z %): it is known that one tablet of Algocalmin® had a content of 500 mg pure sodium metamizole, so $Z = 477,477 / 5 = 95.495\%$ revealed by equation (5). Thus, $Z = 95.495\%$ has expressed sodium metamizole percentage contents in tablet.

Calculated pure sodium metamizole value has represented 95.495% from the officially declared value (500 mg) by the pharmaceutical company and it had an average maximum percentage deviation of only 4.505% from the official pure declared active substance content.

Regression statistic parameters evaluation

Statistic parameters of method linearity which have been determined in Microsoft Excel 2016, were presented in table 4.

Table 4
STATISTIC VALUES OF LINEAR REGRESSION PARAMETERS

Regression Statistics	
Multiple R (Correlation coefficient)	0.999898
R Square R ² (Linear regression coefficient)	0.999795
Adjusted R Square R ²	0.999775
Standard Error (SE)	0.004882
Observations	12

Equation of the regression line was: $y = 0.0234 \cdot x - 0.0031$, or $A_s(\lambda) = 0.0234 \cdot C_s(\mu\text{g/mL}) - 0.0031$. The intercept was (-) 0.0031 and slope : 0.0234. Linear regression coefficient was $R^2 \geq 0.999$ and correlation coefficient $R > 0.999$ were above minimum admissible value (table 3) and were situated within the normal range of values, which demonstrated the linear variation of measured standard solutions absorbances corresponding to their concentrations. Standard error of the regression line (SE) was $SE = 0.004882$ (table 3) had a corresponding, highly low value.

Detection limit (LD) and quantitation limit (LQ); Detection limit was, $LD = 3 \cdot 0.004882 / 0.0234$, thus $LD = 0.626$ µg/mL according to relation (6) and Quantitation limit was assigned with $LQ = 10 \cdot 0.004882 / 0.0234$, so $LQ = 2.086$ µg/mL, from equation (7).

Conclusions

The method used for Visible spectrophotometric analysis of sodium metamizole in tablets marketed under the name Algocalmin Zentiva®, was linear in standard concentrations range 1 Eg/mL -40 Eg/mL; the linear regression coefficient was $R^2 = 0.999795$, $R^2 \geq 0.999$ and correlation coefficient $R = 0.999898$, $R > 0.999$. Standard error of the regression line $SE = 0.004882$, detection limit $LD = 0.626$ Eg/mL and quantitation limit $LQ = 2.086$ Eg/mL were located within the normal range of values.

Visible spectrophotometric (VIS) method used for quantitative analysis of sodium metamizole in tablets has been successfully validated through the complete scrolling of studied stages and could be applied in practice to active substance dosing from different samples.

New methods for detecting the active principles in the various of pharmaceutical forms with spectacular results are *Ordered mesoporous carbon based sensor for Sensitive detection* or *Electrochemical determination* [28-29].

Acknowledgments . This study is simply a scientific research paper that does not aim to confirm or deny the official results of pharmaceutical manufacturing company, nor to cause any image damage.

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